

Biosynthesis of Tuberin from Tyrosine and Glycine

Richard B. Herbert* and Jonathan Mann

Department of Organic Chemistry, The University, Leeds LS2 9JT, U.K.

The *N*-formyl and *O*-methyl groups of tuberin (**8**) are labelled efficiently by [2-¹⁴C]glycine but the *N*-formyl group of (**8**) is not labelled by various other precursors notably [¹⁴C]formate and [*Me*-¹⁴C]methionine [the latter precursor labels the *O*-methyl group of (**8**) very efficiently]; *L*-tyrosine serves as a precursor for the remaining carbons in tuberin (**8**) but its incorporation is not completely intact, and DL-[2-¹⁴C]-*threo*-3-hydroxytyrosine (**5**) is incorporated into (**8**) apparently only *via* glycine.

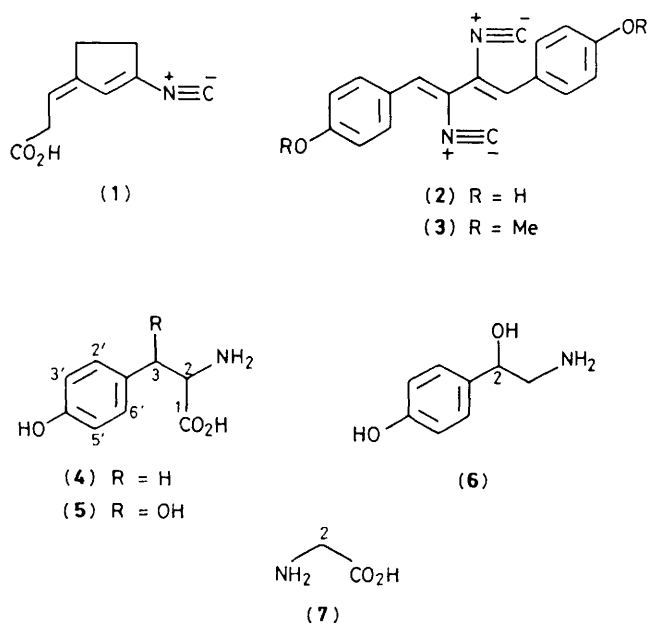
Studies of the biosynthesis of the intriguing microbial isonitriles, (**1**) (*Trichoderma hamatum*)¹ and xanthocillin-X (**2**)

(*Penicillium notatum*),² have shown that the carbon skeleton of each metabolite is elaborated from tyrosine. Critically

though, the origin of the isonitrile atoms has remained unknown. A strikingly similar structure to that of xanthocillin-X (2) is found in the simple metabolite, tuberin (8), which is produced by *Streptomyces amakusaensis*.³ Results reported here define the origins of (8) and have a bearing on the biosynthesis of (2); preliminary and parallel results on the biosynthesis of xanthocillin-X dimethyl ether (3) in *Aspergillus clavatus* indicate that the origin of the isonitrile carbon atoms in (3) may be the same as that of the *N*-formyl group in tuberin (8).⁴

L-[U-¹⁴C]Tyrosine (1.15 μ Ci; 504 mCi mmol⁻¹) was found to be a very efficient precursor for tuberin (8) in *S. amakusaensis* (ATCC 23876) (14–51% incorporation).† L-[1-¹⁴C]Tyrosine (10 μ Ci; 56 mCi mmol⁻¹) acted as a very poor tuberin precursor (0.007%) which shows that the carboxy-group of (4) is lost during biosynthesis just as it is in the biosynthesis of (1)¹ and (2).² A check on the utilization of tyrosine as an intact C₈ unit, in experiments with L-[U-¹⁴C,3',5'-³H₂]- and L-[U-¹⁴C,2',6'-³H₂]-tyrosine [as (4)] (similar weights and activities to other tyrosine samples above), revealed surprisingly that partial degradation of the tyrosine had occurred; the decrease in tritium relative to ¹⁴C was 17–45% (in calculations, here and below, associated with mixed isotopic labelling of tyrosine, allowance was made for the loss of one ninth of the ¹⁴C-activity arising from the loss of the carboxy carbon). That tritium was not being lost by exchange *ortho* to the phenolic hydroxy-group in the [3',5'-³H₂]tyrosine during biosynthesis was confirmed in an experiment with [U-¹⁴C,3',5'-²H₂]tyrosine [as (4)] (84% ²H₂ species, 10 mg; 0.8 μ Ci). The tuberin obtained was shown, by mass spectrometry, to contain a significant amount of dideuterated species but \leq 1% of monodeuterated species.

The specific incorporation‡ of ¹⁴C-label into (8) was higher (10.9%) than that of deuterium (7.5%) but the specific incorporation measured on the degradation product (10) was very similar (7.8%) to the specific incorporation of deuterium

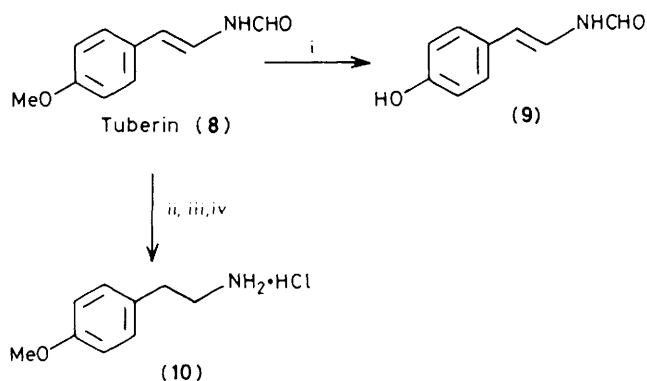


into (8). It follows that the tyrosine fed had undergone partial degradation, and 15% of the ¹⁴C-label from (4) was located in the *N*-formyl group of (8); alternative degradation of (8) to (9) was without loss of radioactivity (degradations were carried out on 2 samples of tuberin obtained from tyrosine in different experiments) which showed that no labelling of the *O*-methyl group of tuberin (8) by tyrosine had occurred.

[Me-¹⁴C]Methionine (18.3 μ Ci; 56 mCi mmol⁻¹) served as an efficient precursor for tuberin (8) (21% incorporation). Essentially all the radioactivity was lost on conversion of the tuberin into (9). Thus the methyl group of methionine is a source of the *O*-methyl group in (8) but not of the *N*-formyl group. Neither [¹⁴C]formate (3.2 μ Ci; 59 mCi mmol⁻¹) (0.02% incorporation) nor [2-¹⁴C]acetic acid (5 μ Ci; 59 mCi mmol⁻¹) (0.01%) or [U-¹⁴C]pyruvic acid (4 μ Ci; 16.6 mCi mmol⁻¹) (0.02%) were significantly incorporated into (8). Similar results have been obtained for (2).²

We considered that octopamine (6) or 3-hydroxytyrosine (5) might be involved as intermediates in tuberin biosynthesis. [2-³H]Octopamine (2 mg; 25 μ Ci) was not significantly incorporated (0.05%) but DL-[2-¹⁴C,3',5'-²H₂]-threo-3-hydroxytyrosine [as (5)] (synthesis following ref. 5) (11 mg; 3.2 μ Ci) served as a satisfactory precursor albeit at a lower level than tyrosine (1.0%). However, the tuberin (8) was devoid of deuterium (0.9% enrichment required for intact incorporation) and 60% of the ¹⁴C-label was lost on degradation to (10). Thus (5) serves only as a C₁ source labelling the *N*-formyl group and, probably, the *O*-methyl group. It is known that glycine can serve as a C₁ source through methylene tetrahydrofolate⁶ and has been noted to do so recently in the biosynthesis of another *Streptomyces* metabolite, naphthyrindinomycin.⁷ It is suggested, and supported by other results below, that labelled 3-hydroxytyrosine (5) is degraded by a reverse-aldol reaction to give [2-¹⁴C]glycine [as (7)] which labels the C₁ units in (8). Since (5) is stable in aqueous solution under the conditions of the feeding experiment the putative reverse-aldol reaction is likely to be enzyme-mediated and the sequence (4) \rightarrow (5) \rightarrow (7) may account for the partial degradation observed on incorporation of tyrosine (4) into tuberin (8).§ However, the failure of [U-¹⁴C]tyrosine, unlike labelled (5) and (7), to label the *O*-methyl group of (8) argues against this.

[2-¹⁴C]Glycine (10 μ Ci; 49.5 mCi mmol⁻¹) was found to be a satisfactory precursor for tuberin (8) (1.6–3.2% incorpora-



Reagents: i, BBr₃; ii, H₂/Pd-C;¹⁰ iii, 1 M aq. NaOH;¹⁰ iv, HCl.¹⁰

† % Incorporation is: $100 \times \frac{\text{[total radioactivity in (8) isolated]}}{\text{[total radioactivity in precursor]}}$.

‡ Specific incorporation as used here for ¹⁴C and ²H is: $100 \times \frac{\{\text{amount of label per mmol of (8) [or (10)]}\}}{\{\text{amount of label per mmol of precursor}\}}$.

§ It should be noted that the nitrogen atom and C-2 of tyrosine serve, as does alternatively glycine, as the source for a C₁-N unit in thiamine biosynthesis (refs. 8,9), but here the remainder of the tyrosine skeleton is converted into *p*-hydroxybenzyl alcohol (ref. 9) without intervention of the reverse-aldol product, *p*-hydroxyphenylacetaldehyde.

tion). The results of degradation to (9) and (10) showed that 56% of the activity was in the *N*-formyl group and 38% in the *O*-methyl group of tuberin (8), a very similar distribution to that deduced for the [2-¹⁴C]-3-hydroxytyrosine. We conclude that C-2 of glycine is specifically the source for the C₁-units in tuberin probably *via* methylenetetrahydrofolate. It is possible that glycine may provide the *N*-formyl group of (8) instead by way of glyoxylic acid, its transamination product, but the close similarity in the labelling levels of the C₁ units in (8) argues against this.

Glycine can also serve as a precursor for (3).⁴ Work is now in hand to define closely the origins of the isonitrile (3), the detailed biosynthetic pathways to (8) and (3), and the relationship of these pathways to one another.

We thank the University of Leeds for a Scholarship (to J. M.) and for financial support from its Research Fund.

Received, 10th June 1983; Com. 768

References

- 1 R. J. Parry and H. P. Buu, *Tetrahedron Lett.*, 1982, **23**, 1435; J. E. Baldwin, A. E. Derome, L. D. Field, P. T. Gallagher, A. A. Taha, V. Thaller, D. Brewer, and A. Taylor, *J. Chem. Soc., Chem. Commun.*, 1981, 1227.
- 2 H. Achenbach and F. König, *Chem. Ber.*, 1972, **105**, 784; H. Achenbach and H. Grisebach, *Z. Naturforsch., Teil B*, 1965, **20**, 137.
- 3 K. Ohkuma, K. Anzai, and S. Suzuki, *J. Antibiot.*, 1962, **15**, 115.
- 4 R. B. Herbert and J. Mann, unpublished results.
- 5 W. A. Bolhofer, *J. Am. Chem. Soc.*, 1954, **76**, 1322.
- 6 See, *e.g.*, D. E. Metzler, 'Biochemistry,' Academic Press, New York, 1977, p. 837; E. B. Newman and B. Magasanik, *Biochim. Biophys. Acta*, 1963, **78**, 437; H. Kochi and G. Kikuchi, *J. Biochem. (Tokyo)*, 1974, **75**, 1113.
- 7 M. J. Zmijewski, Jr., M. Mikolajczak, V. Viswanatha, and V. J. Hruby, *J. Am. Chem. Soc.*, 1982, **104**, 4969.
- 8 G. M. Brown and M. Williamson, *Adv. Enzymol.*, 1982, **53**, 345.
- 9 R. H. White, *Biochim. Biophys. Acta*, 1979, **583**, 55.
- 10 K. Anzai, *J. Antibiot.*, 1962, **15**, 117.